

Physics 330, Spring 2009
Exam 2
100 points

1. **Problem 1 (30 points)** Consider the hyperfine splitting in the Hydrogen atom. The transition from the higher energy state to the lower energy state gives a 21 cm wavelength photon.

- (a) (10 points) At the temperature of the cosmic microwave background, 2.7 K, what is the population ratio of higher energy state to the lower energy state?
- (b) (15 points) What is the effective magnetic field that produces the energy level splitting? Express your answer in terms of the wavelength given above, fundamental constants and the Bohr magneton.
- (c) (5 points) What is the origin of this effective B-field?

2. **Problem 2 (30 points)** The Li atom has $Z = 3$.

- (a) (20 points) Neglecting any fine-structure level splitting (assume energy levels are determined only by n), give all possible j -values for the first excited state of the neutral Li atom.
- (b) (10 points) Give an example of a transition from the first excited state to the ground state that is forbidden. Explain what selection rule is being violated.

3. **Problem 3 (20 points)**

- (a) (10 points) Sketch the Fermi-Dirac distribution, $f_{FD}(E)$, versus E at $T = 0$ K.
- (b) (10 points) What is Bose-Einstein condensation?

4. **Problem 4 (20 points)** KBr has a equilibrium spacing of $r_0 = 0.3$ nm and a vibrational frequency of $f = 7 \times 10^{12}$ Hz. Calculate the spacing between the two lowest rotational energy levels. The masses of K and Br are $m=39$ u and $m=80$ u, respectively. $u=1.7 \times 10^{-27}$ Kg.



$$T = 2.7 \text{ K}$$

$$hc = 1240 \text{ eV}\cdot\text{nm}$$

$$= 1.24 \times 10^{-4} \text{ eV}\cdot\text{cm}$$

a.)

$$E = \frac{hc}{\lambda}$$

$$\frac{N_2}{N_1} = \frac{e^{-E_2/KT}}{e^{-E_1/KT}} = e^{-(E_2 - E_1)/KT}$$

$$\frac{N_2}{N_1} = e^{-\left(\frac{hc}{\lambda}\right)/KT}$$

$$\frac{N_2}{N_1} = e^{-\left(\frac{hc}{21 \text{ cm}}\right)/K(2.7 \text{ K})}$$

estimate :

at $T = 300 \text{ K}$, $kT \sim \frac{1}{40} \text{ eV}$

at $T = 2.7 \text{ K}$, $kT \sim \frac{1}{4000} \text{ eV} \sim 0.00026 \text{ eV}$

$$\frac{hc}{\lambda} = \frac{1.24 \times 10^{-4} \text{ eV}\cdot\text{cm}}{21 \text{ cm}} \sim 6 \times 10^{-5} \text{ eV}$$

$$\left(\frac{hc}{\lambda}\right)/KT \approx \frac{6 \times 10^{-5}}{2.1 \times 10^{-4}} \sim 0.3$$

$e^{-0.3} \leftarrow$ significant population of N_2

b.)

$$\Delta U = 2\mu_B B_z$$

for the electron, $\mu_z = -m_s g_s \mu_B$, $m_s = \pm 1$

$$\Delta U = 2\mu_B B$$

$g_s \sim 2$

$$B = \frac{\Delta U}{2\mu_B} = \frac{hc}{2\lambda\mu_B} = \frac{hc}{2(21 \text{ cm})\mu_B}$$



c.) proton spin

(2.)

$Z=3$. Ground state : $1s^2 2s^1$

a.) first excited state is outer electron
in $n=3$ shell.

inner electrons combine to give $J=0$, so
total angular momentum is from outer electron,

for $n=3$, $l=0, 1, 2$
 $s = \frac{1}{2}$

$j = \frac{1}{2}, \frac{3}{2}, \frac{5}{2}$

b.)

$3S_{\frac{1}{2}} \rightarrow 2S_{\frac{1}{2}}$

$\Delta l = \pm 1$

$3P_{\frac{5}{2}} \rightarrow 2S_{\frac{1}{2}}$

$\Delta j = \pm 1, 0$

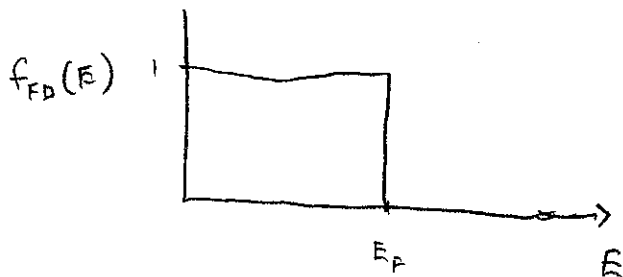
$3P_{\frac{3}{2}} \rightarrow 2P_{\frac{3}{2}}$

$\Delta l = \pm 1$ (although not to
ground state)

3.

a.)

$$f_{FD} = \frac{1}{e^{(E-E_F)/kT} + 1}$$



b.) For Bose particles (bosons) there is no restriction on the number of particles allowed in a certain quantum state.

As $T \rightarrow 0$, all particles "condense" in to the ground state.

4.

$$r_0 = 0.3 \text{ nm}$$

$$f = 7 \times 10^{12} \text{ Hz}$$

$$m_K = 39 \text{ u}$$

$$m_{Br} = 80 \text{ u}$$

$$\mu = 1.7 \times 10^{-27} \text{ kg}$$

$$E_r = \frac{L^2}{2I}$$

$$I = \mu r_0^2 = \frac{m_K m_{Br}}{m_K + m_{Br}} r_0^2$$
$$L^2 = l(l+1) \hbar^2$$

$$E_r = \frac{L^2}{2r_0^2 \left(\frac{m_K m_{Br}}{m_K + m_{Br}} \right)} = \frac{l(l+1) \hbar^2}{2r_0^2} \cdot \frac{m_K + m_{Br}}{m_K m_{Br}}$$

Lowest two rotation energy levels are $l=0, 1$

$$\text{so } \Delta E_r = \frac{\hbar^2}{2r_0^2} \left(\frac{m_K + m_{Br}}{m_K m_{Br}} \right) = \frac{(1.05 \times 10^{-34} \text{ J}\cdot\text{s})^2}{2(3 \times 10^{-10} \text{ m})^2} \left(\frac{119}{80 \cdot 39} \right) \left(\frac{1}{1.7 \times 10^{-27} \text{ kg}} \right)$$
$$= 1.4 \times 10^{-24} \text{ J}$$
$$= 8.6 \times 10^{-6} \text{ eV}$$