

Physics 330 HW # 7

(1.) H-atom in 3D state

a.) possible values of  $j$ .

$$l = 2$$

$$s = \frac{1}{2}$$

$$j = \frac{3}{2}, \frac{5}{2}$$

b.)  $J = \sqrt{j(j+1)} \hbar$

$$\frac{3}{2} : \sqrt{\frac{3}{2} \left(\frac{5}{2}\right)} \hbar = \frac{\sqrt{15}}{2} \hbar$$

$$\frac{5}{2} : \sqrt{\frac{5}{2} \left(\frac{7}{2}\right)} \hbar = \frac{\sqrt{35}}{2} \hbar$$

c.)  $J_z = m_z \hbar$  ,  $m_z = -j, -j+1, \dots, j$

$$J_z = -\frac{5}{2} \hbar, -\frac{3}{2} \hbar, -\frac{1}{2} \hbar, \frac{1}{2} \hbar, \frac{3}{2} \hbar, \frac{5}{2} \hbar$$

2.)

$$l=1, s=\frac{1}{2}$$

a.) possible values of  $l$  corresponds to  $\vec{L} = \vec{L}_1 + \vec{L}_2$

$$l = 0, 1, 2$$

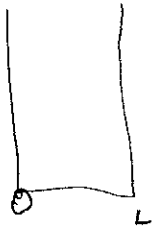
b.)  $s = 0, 1$

c.)  $\vec{J} = \vec{L} + \vec{S}$  ,  $j = 0, 1, 2, 3$

d.)  $j_1, j_2 = \frac{1}{2}, \frac{3}{2}$

e.)  $0, 1, 2, 3$

3.



two neutrons

$$L = 2 \times 10^{-15} \text{ m}$$

Can't both be in  $n=1$  state

$$\psi_1(x_1) = \sqrt{\frac{2}{L}} \sin \frac{\pi x_1}{L}$$

$$\psi_2(x_2) = \sqrt{\frac{2}{L}} \sin \frac{2\pi x_2}{L}$$

make anti-symmetric wave function

$$\Psi(x_1, x_2) = \frac{1}{\sqrt{2}} \left[ \psi_1(x_1) \psi_2(x_2) - \psi_2(x_1) \psi_1(x_2) \right]$$

Schrödinger eq.

$$-\frac{\hbar^2}{2m} \left( \nabla_1^2 + \nabla_2^2 \right) \Psi(x_1, x_2) = E \Psi(x_1, x_2)$$

$$\text{using } E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$

$$\frac{1}{\sqrt{2}} \left[ E_1 \psi_1(x_1) \psi_2(x_2) - E_2 \psi_2(x_1) \psi_1(x_2) \right]$$

$$+ \frac{1}{\sqrt{2}} \left[ E_2 \psi_1(x_1) \psi_2(x_2) - E_1 \psi_2(x_1) \psi_1(x_2) \right] = E_{\text{tot}} \frac{1}{\sqrt{2}} \left[ \psi_1(x_1) \psi_2(x_2) - \psi_2(x_1) \psi_1(x_2) \right]$$

$$= (E_1 + E_2) \Psi(x_1, x_2) = E_{\text{tot}} \Psi(x_1, x_2)$$

$$E_{\text{tot}} = 4.1 \times 10^{-11} \text{ J} = 25 \text{ MeV}$$

$$E_{\text{tot}} = E_1 + E_2 = \frac{5 \pi^2 \hbar^2}{2mL^2} = \frac{5 \pi^2 (1.055 \times 10^{-34} \text{ J}\cdot\text{s})^2}{2 \cdot (1.67 \times 10^{-27} \text{ kg}) (2.0 \times 10^{-15} \text{ m})^2}$$

as expected  $\uparrow$

4.

a.)  $z = 6$

$$1s^2 2s^2 2p^2$$

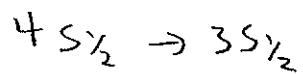
b.)  $z = 8$

$$1s^2 2s^2 2p^4$$

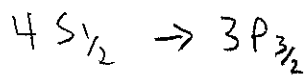
c.)  $z = 18$

$$1s^2 2s^2 2p^6 3s^2 3p^6$$

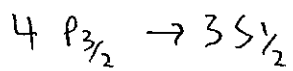
5.



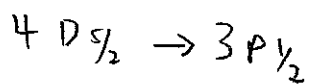
forbidden  $\Delta l \neq \pm 1$



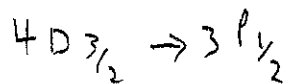
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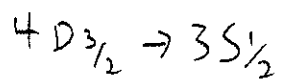
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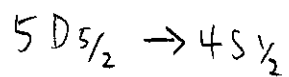
forbidden  $\Delta J \neq \pm 1, 0$ , also  $\Delta m_s \neq 0$



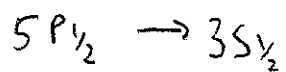
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forbidden  $\Delta l \neq \pm 1$



forbidden  $\Delta J \neq \pm 1, 0$  also  $\Delta m_s \neq 0$



OK.

(6)

$$a.) \quad E = \frac{hc}{\lambda}$$

$$4P_{3/2} \quad E = \frac{1240 \text{ eV} \cdot \text{nm}}{766.41 \text{ nm}} = 1.618 \text{ eV}$$

$$4P_{1/2} \quad E = \frac{1240 \text{ eV} \cdot \text{nm}}{769.90} = 1.611 \text{ eV}$$

$$b.) \quad \Delta E = 7.334 \text{ meV}$$

$$c.) \quad \mu_z = -m_s g_s \mu_B$$

$$\Delta E = 2U$$

$$U = -\mu_z B_z$$

$$\Delta E = 2B_z m_s g_s \mu_B$$

$$\approx 2B_z \mu_B$$

$$B_z \approx \frac{\Delta E}{2\mu_B} = \frac{(7.334 \times 10^{-3} \text{ eV})}{2(5.79 \times 10^{-5} \text{ eV/T})}$$

$$B \approx 63.3 \text{ T}$$