

Wave Packets

$$\Psi(x) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} dk g(k) e^{+ikx}$$

$$g(k) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} dx \Psi(x) e^{-ikx}$$

$$\frac{dp}{dx} = |\Psi(x)|^2 - \text{probability per unit distance of finding a particle}$$

A precisely localized particle

$$\frac{dp}{dx} = |\Psi(x)|^2 = \lim_{L \rightarrow 0} \left[\frac{1}{\sqrt{2\pi L^2}} e^{-\frac{(x-a)^2}{2L^2}} \right] \equiv \delta(x-a)$$

Dirac Delta function

$$\int_{-\infty}^{\infty} f(x) \delta(x-a) dx = f(a)$$

$$g(k) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} dx \delta(x-a) e^{-ikx} = \frac{1}{\sqrt{2\pi}} e^{-ika}$$

$$|g(k)|^2 = \frac{1}{2\pi} \sim \text{needs all waves from } k = -\infty \text{ to } k = +\infty \text{ to make zero width pulse}$$

Gaussian wave packets and the uncertainty principle

$$\Psi(x) = C e^{-\frac{x^2}{2L^2}}$$

$\sigma_x = L$ - standard deviation of Gaussian distribution

$$g(k) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} dx C e^{-x^2/2L^2} e^{-ikx}$$

$$= \frac{C}{\sqrt{2\pi}} \sqrt{\frac{\pi}{2L^2}} e^{-k^2 L^2 / 2}$$

$$\sigma_k = \frac{\hbar}{L}$$

$$\sigma_x \sigma_k = 1$$

Probability distributions are $|\Psi(x)|^2$, $|g(k)|^2$

$$|\Psi(x)|^2 = C^2 e^{-x^2/L^2}$$

$$\sigma_x = \frac{L}{\sqrt{2}}$$

$$\sigma_k = \frac{1}{\sqrt{2L}}$$

$$\sigma_x \sigma_k = \frac{1}{2} \quad P = \frac{h}{\lambda} = \frac{\hbar k}{2}$$

$$\sigma_x \sigma_p = \frac{\hbar}{2}$$

Gaussian wave packets give the smallest

$\sigma_x \sigma_p$ product. so...

$$\Delta x \Delta p \geq \frac{\hbar}{2}$$

particles confined to finite space
have a minimum energy

$$\langle E_k \rangle = \frac{\langle p \rangle^2}{2m}$$

$$\Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2}$$

\uparrow goes to zero since \pm equally likely

$$(\Delta p)^2 = \langle p^2 \rangle$$

$$\langle E_k \rangle = \frac{(\Delta p)^2}{2m} \sim \text{sometimes called 'zero point energy'}$$

In class problem:

Estimate the energy - ~~Kinetic energy~~ of
an electron confined to the size of an
atom $\sim 0.1 \text{ nm}$?

Compare this to $E_k = \frac{Ke^2}{r} \sim$ the kinetic energy
of an electron
used in the Bohr model.

H W #2

5.2 , 5.5 5.14 5.19

Due Wed. Jan 31